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Fractional distillation. The solution is heated to the temperature of the substance with the lowest boiling point.; This substance will rise and evaporate first, and vapours will pass through a condenser, where they cool and condense, turning into a liquid that will be collected in a beaker.; For water and ethanol: ethanol has a boiling point of  $78^{\circ}\text{C}$  and water of  $100^{\circ}\text{C}$ .

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Practical WAEC GCE Chemistry 2020  
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Mark Scheme of Cambridge IGCSE  
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answers and calculations has been detailed  
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