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4 Basis: An atom, ion, molecule or group  
of atoms is called a point and each point is  
connected with the other points to form a

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**Solutions** specific body. The basis for each point should be: 1 - Symmetrical parts in terms of structure, arrangement and direction. 2.

*Solid State Physics Chapter 1: Crystal Structure*

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*Page 12/35*

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Asst. Prof. Dr. Tagreed Muslim 4 Basis:

An atom, ion, molecule or group of atoms is called a point and each point is connected with the other points to form a specific body. The basis for each point should be: 1 - Symmetrical parts in terms of

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If NaCl is doped with 10 – 4 mol % of Sr

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Cl<sub>2</sub>, the concentration of cation vacancies  
will be ( $N_A = 6.02 \times 10^{23} \text{ mol}^{-1}$ )  
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state but ionic solids do so only in molten  
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electricity through electrons and ionic substances through ions. Metallic solids are malleable and ductile while ionic solids are hard and brittle.

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chapter (Hindi ...*

Ans: It is a covalent or network solid. 1.8.

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**Solutions**  
Why are ionic solids conducting in the molten state and not in the solid state ?

Ans: In the ionic solids, the electrical conductivity is due to the movement of the ions. Since the ionic mobility is negligible in the solid state, these are non-conducting in this state.

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